

## Formation of Substituted 1, 2, 4- triazole and evaluation of antifungal activities with ADMET study

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### Abstract

A series of 1,2,4-triazole derivatives were synthesized from ethyl 4-hydroxybenzoate with various substituents, and the data came from spectroscopy, infrared spectroscopy (FT-IR), nuclear magnetic resonance (<sup>1</sup>H NMR), (<sup>13</sup>C NMR ) resonance. The compounds' identities were determined using mass spectrometry. The compounds' antifungal efficacy was tested on two types of plant fungi, and the results show that some of them are involved in promising activities, which were also studied using the computer for the prepared compounds. (B3, B4 ,B5)

**Keywords:** *triazole, derivatives, ADMET, Rhizoctonia solani, Fusarium spp.*

### INTRODUCTION

For decades, fungicides have been widely used to control fungal-caused plant diseases that threaten human health and crop production. Seed rot is caused by a fungus group that includes *Fusarium*, *Rhizoctonia*, and *Sclerotium* species, and it infects plants at the seedling stage. These pathogens have the ability to infect a wide range of plants. Because they live in wet soils, seedling pathogens frequently cause damping-off symptoms [1, 2]. Triazole compounds have been widely used to prevent and control fungal diseases of many crops (fruits, vegetables, nuts, grains, and seeds) since the 1980s, quickly spreading throughout the European market and becoming the most effective type of fungicide. Triazole compounds are significant due to their high antifungal activity, low resistance risk, and long-term

stability in soil and water. Their strategy is based on preventing fungal ergosterol biosynthesis as well as steroid demethylation [3]. Triazole derivatives have a wide range of biological effects [4].antibacterial [5], fungicidal [6, 7], antiviral [8], antiproliferative [9, 10], antioxidant [11,12] cytotoxic and antitumor [13], antileishmanial [14], anti-tubercular [15], anti-inflammatory and analgesic [16], anticonvulsant [17], anti-obesity [18], antimalarial [19], and anticancer [20] properties. Antifungal medications include fluconazole, itraconazole, voriconazole, and ketoconazole. Nazole is a triazole derivative that has been approved for use as an antifungal agent. [21,22].

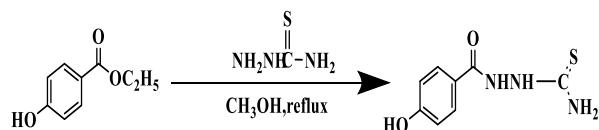
### Experimental

All starting materials and solvents were obtained from BDH and Sigma Alderch and

were used without further modification or purification. All starting materials and solvents were obtained from BDH and Sigma-Aldrich Companies and were used without modification or purification. Melting points were determined uncorrected in an open capillary tube. The FT-IR Spectrophotometer, Shimadzu IR -8400s spectrometer was used to record FT-IR infrared spectra. The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra were obtained using a 400 MHz Bruker Biospin GmbH. Chemical shifts are reported as values in parts per million (ppm) using DMSO- $d_6$  as a solvent while mass spectroscopy was recorded. Predictions on a GC-Shimadzu QP-2010 plus, The completion of the reactions was checked by TLC using hexane and ethyl acetate (6:4), ethanol and cyclohexane (4:1), ethyl acetate and petroleum ether (5:5) and iodine vapors as a detecting reagent, ADMET and in-silico predictions the proposed compounds' structures are converted into a Simplified Molecular-Input Line-Entry system using the admetSAR 2.0 site. (SMILE). The same site then submitted chemical compounds like AlogP, H-Bond Acceptor, H-Bond Donor, Rotatable Bonds, Molecular Weight, and Applicability Domain to predict pharmacokinetics in silico. Using the same tool, human oral bioavailability and intestinal absorption were also predicted.

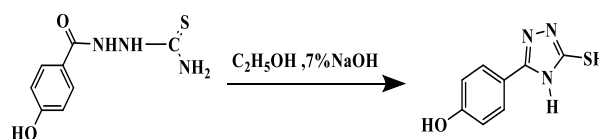
### 1: Synthesis of (4-hydroxybenzoyl) hydrazine-1-carbothioamide (B1)

Ethyl 4-hydroxybenzoate (0.01 mol, 1.6618 g) was dissolved in methanol (30 mL), and then thiosemicarbazide (0.01 mol, 0.9114 g) was dissolved in (30 mL) of methanol. The reaction mixture was heated under a reflux flow for 8 hours, then allowed to cool, after which the resulting precipitate was collected to be re-purified from absolute ethanol [23].



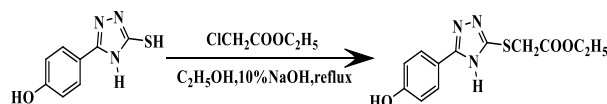
### 2: Synthesis of 4-(5-mercpto-4H-1, 2, 4-triazol-3-yl) phenol (B2)

The B1 (0.007 mol, 1.477 g) was dissolved in (15 mL) of absolute ethanol. The mixture was refluxed for (3-4) hours with (12 mL) of a 7% aqueous sodium hydroxide solution. After the end of the reaction, it was left to cool, and after that, the filtrate was acidified using dilute hydrochloric acid. The precipitate was filtered and washed several times with water. and recrystallization from absolute ethanol [24].



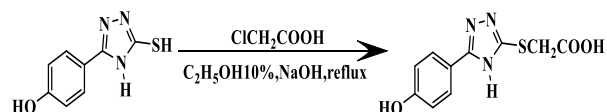
### 3: Synthesis of ethyl 2-((5-(4-hydroxyphenyl)-4H-1, 2, 4-triazol-3-yl) thio) acetate (B3)

After dissolving the B2 (0.01 mol, 1.93 g) in (25 mL) of absolute ethanol and stirring with (5 mL) of a 10% aqueous sodium hydroxide solution, and adding the ethyl chloroacetate (0.01 mol, 1.070 mL) dissolved in (5 mL) of absolute ethanol. The mixture was refluxed for three hours. The solution, after it had cooled, was acidified with concentrated hydrochloric acid. The precipitate was filtered and crystallized from 50% absolute ethanol [25].



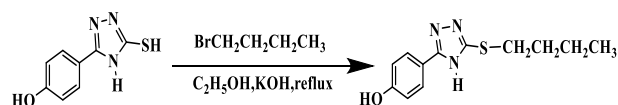
#### 4: Synthesis of 2-((5-(4-hydroxyphenyl)-4H-1,2,4-triazol-3-yl) thio) acetic acid.(B4)

After dissolving the B2 (0.005 mol, 0.965 g) in (20 mL) of absolute ethanol and stirring with (10mL) of 10% aqueous sodium hydroxide solution, and then adding the chloroacetic acid (0.005 mol, 0.472 g) dissolved in (5 mL) of absolute ethanol, the mixture was refluxed for three hours. The solution, after it had cooled, was acidified with concentrated hydrochloric acid. The precipitate was filtered and crystallized from absolute ethanol [26].



#### 5: Synthesis of 4-(5-(butylthio)-4H-1, 2, 4-triazol-3-yl) phenol. (B5)

After dissolving the B2 (0.002 mol, 0.386 g) in (15 mL) of absolute ethanol, it was placed in a round-bottom flask (50 mL) with (0.002mol, 0.1122 g) Potassium hydroxide, and after continuous stirring, (0.002 mol, 0.215mL) butyl bromide dissolved in (5 mL) absolute ethanol. The solution after it had cooled then collected the precipitate and recrystallization from absolute ethanol [27].



#### 6: Synthesis of 3-(4-hydroxyphenyl) thiazolo [2,3-c][1,2,4]triazole-6(5H)-one .(B6)

A mixture B2 (0.004 mol, 0.772 g) and chloroacetyl chloride (0.004 mol, 0.318 mL) in dry dioxins (10 mL) was left to stand at room temperature over night. The precipitated solid was separated and recrystallized from benzene. [28]



#### Antifungal activity testing

The fungi used in the study were obtained from the Department of Plant Protection Research, the Department of Agricultural Research, *Rhizoctonia solani*, and *Fusarium* spp, and the isolates were activated at 25 °C for 5 days for both fungi. Potato dextrose agar media were made by dissolving 39 g per liter of distilled water. The compounds were prepared as a stock solution at 100 ppm using distilled water as a solvent, and the required concentrations (75, 50, and 25 ppm) were prepared by diluting the stock solution with the same solvent. Petri plates, pipettes, spoons, test tubes, needles, media, distilled water, and a cork borer were sterilized in an autoclave at 121°C and 15 psi for 15 minutes. Was poured into (9cm) diameter sterile glass dishes, and compound concentrations (100, 75, 50, and 25 ppm) and medium were added in three replicates for each concentration. (100, 75, 50, and 25 ppm) Only one dish was tested for fungi as a first control, and a plate with 1 mL/L beltanol pesticide was added as a second comparison. It was allowed to harden before being inoculated by inserting the divided colony through a cork puncture in the center of the dish, and it was then transferred to a 25-degree Celsius incubator for 5 days. [29] The percentage of mycelial growth inhibition was calculated by using a millimeter scale and calculating the fungus's growth diameter, followed by calculating the : (%) mycelial growth inhibition=  $A-B / A \times 100$  where, A = Colony growth of the fungal control plate. B= Colony growth of the fungal in treated plate.

## Results and Discussion

A new series of 1, 2, 4-triazole derivatives (B3to B6) were synthesized and yielded good results. All physical parameters and mass

spectrum values in the Table (1).The prepared compounds were identified by spectrophotometric measurements, infrared spectrometers,  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR, the values are shown in the tables (2, 3, 4,).

**Table (1): lists some of the physical properties and yield percentages and m/z for compounds (B1-B6)**

Comp. Symbol	M.F	M.w	m.p.°C	Yield (%)	Color	R <sub>f</sub> Value	m/z
B <sub>1</sub>	C <sub>8</sub> H <sub>9</sub> N <sub>3</sub> O <sub>2</sub> S	211	178-180	74	colorless	0.6	-
B <sub>2</sub>	C <sub>8</sub> H <sub>7</sub> N <sub>3</sub> OS	193	198-200	70	colorless	0.58	-
B <sub>3</sub>	C <sub>12</sub> H <sub>13</sub> N <sub>3</sub> O <sub>3</sub> S	279	150-152	50	colorless	0.51	279
B <sub>4</sub>	C <sub>10</sub> H <sub>9</sub> N <sub>3</sub> O <sub>3</sub> S	251	238-240	52	colorless	0.73	251
B <sub>5</sub>	C <sub>12</sub> H <sub>15</sub> N <sub>3</sub> OS	249	198-200	40	Colorless	0.49	249
B <sub>6</sub>	C <sub>10</sub> H <sub>7</sub> N <sub>3</sub> O <sub>2</sub> S	233	115-118	55	Colorless	0.54	233

**Table (2): Absorption values the (FT-IR) spectrum of compounds (B1-B6)**

Comp. symbol	O-Hst. cm <sup>-1</sup>	N-Hst. cm <sup>-1</sup>	C-Hst. (Ar.)cm <sup>-1</sup>	C-Hst. (al.)cm <sup>-1</sup>	C=Ost. cm <sup>-1</sup>	C=Cst. cm <sup>-1</sup>	Other group
B <sub>1</sub>	3169	3308 3255 3357	3030	-	1641	1529 -1574	C=S 1156 C-N 1277
B <sub>2</sub>	3173	3356 3256	-	-	-	1595	S-H 2564.02 C=N1686 C-N 1275
B <sub>3</sub>	3199	3260 3323	3033	2945 2902	1716	-	C-N1329 C-S 669
B <sub>4</sub>	3310	3260	3034	2988 2971	1682	1554.98	C=N1595 C-N 1250 C-S 692
B <sub>5</sub>	3167	3253	-	2978 2903	-	1614.90	C=N 1640 C-Hben1400-1449 C-N1225
B <sub>6</sub>	3191	-	-	2978.5 2816	1667	-	C=N1640 C-N 1280 C-S 705

Table (3): Chemical shift values (<sup>1</sup>H NMR) of compounds (B2 –B5)

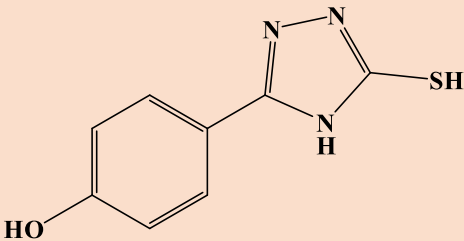
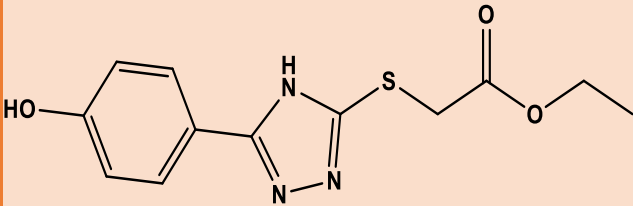
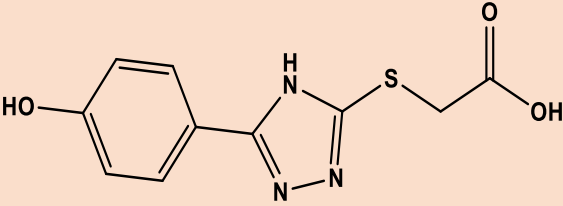
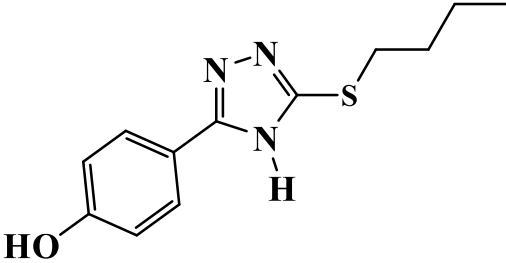
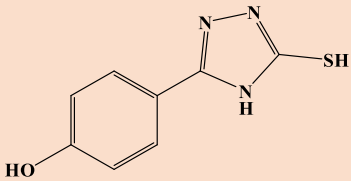
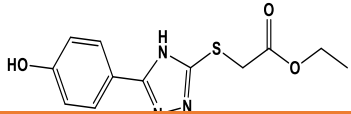
Comp. symbol	Structure	Chemical Shift(ppm)	No. of Protons	Type of signle	group
B <sub>2</sub>		7.02-8.04	4	d,d	Ar. protons
		10.09	1	s	OH-
		11.53	1	s	NH-
		13.59	1	s	SH-
B <sub>3</sub>		1.10-1.13	3	t	-CH <sub>3</sub>
		3.98	2	s	CH <sub>2</sub> -C=O
		4.02-4.06	2	m	O-CH <sub>2</sub>
		7.09-7.95	4	d,d	Ar. protons
		9.84	1	s	-OH
		11.69	1	s	-NH
B <sub>4</sub>		3.98	2	s	CH <sub>2</sub> -C=O
		7.28-8.04	4	d,d	Ar. protons
		10.22	1	s	-OH
		11.18	1	s	-NH
		12.62	1	s	OHCarbox.
B <sub>5</sub>		0.83-0.86	3	t	-CH <sub>3</sub>
		1.27-1.34	2	m	-CH <sub>2</sub>
		1.58-1.63	2	m	-CH <sub>2</sub>
		3.00-3.03	2	t	S-CH <sub>2</sub>
		6.91-7.84	4	d,d	Ar. protons
		9.75	1	s	-OH
		11.43	1	s	-NH

Table (4): Chemical shift values (<sup>13</sup>C NMR) for compounds(B2- B5)

Comp. symbol	Structure	Chemical Shift (δ ppm)	group
B <sub>2</sub>		116.22-124.33-128.63	Ar. Carbons
		144.31-144.63	-C- Hetero ring
		145.48	C-OH
B <sub>3</sub>		15.81	CH <sub>3</sub>
		59.71	-CH <sub>2</sub> -

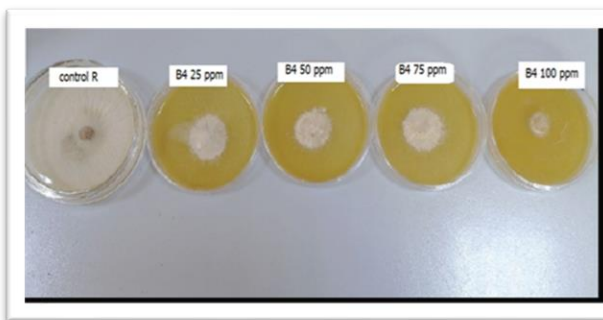


B <sub>4</sub>	4.35	51.67	3.75	58.33	3.55	60.56	1.80	80.00
B <sub>5</sub>	3.50	61.11	2.63	70.81	2.50	72.22	2.15	76.11
B <sub>6</sub>	9.00	0.00	9.00	0.00	9.00	0.00	9.00	0.00
Beltanol	0.00							
Control	9.00							
LSD 5%	Treatments		concentration					
	0.54**	6.10**		1.69**				

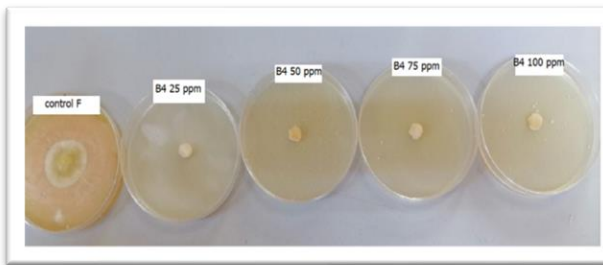
**Table (6): The inhibition and colony dimeter for compounds on the fusarium spp**

Treatments	Concentration							
	25(ppm)		50(ppm)		75(ppm)		100(ppm)	
	CD	PGI%	CD	PGI%	CD	PGI%	CD	PGI%
B <sub>3</sub>	0.00	100.00	0.00	100.00	0.00	100.00	0.00	100.00
B <sub>4</sub>	1.00	88.89	1.00	88.89	1.00	88.89	1.00	88.89
B <sub>5</sub>	1.00	88.89	1.00	88.89	1.00	88.89	1.00	88.89
B <sub>6</sub>	1.50	83.33	2.50	72.22	2.88	68.04	3.13	65.22
Beltanol	0.00							
Control	9.00							
LSD 5%	Treatments		concentration					
	0.24**	2.76**		0.77**				

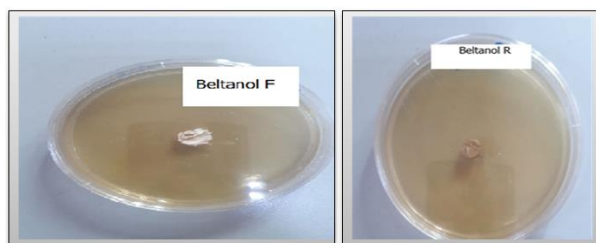
**Figure (1): Antifungal activity of prepared compound (B4) against Rhizoctonia solani**



**Figure (2): Antifungal activity of prepared compound (B4) against Fusarium spp**



**Figure (3): Antifungal activity of Beltanol pesticide against Fusarium spp and Rhizoctonia solani**



#### In-silico & ADMET Prediction Models

Computer-aided by in silico software, which relates to the most well-known biological terms, in vivo and in vitro, Absorption, distribution, metabolism, excretion, and toxicity are the five criteria that make up ADMET for the compound. Molecular Mass: The molecular masses of pharmaceutical drugs must not exceed 500 daltons under Lipinski's rule. All prepared triazole derivatives comply with B3-B5 values between (249.34 and

279.32). Lipophilicity (Octanol/water split coefficient): In a two-stage octanol/water system, the octanol/water fraction ( $\log P$ ) illustrates the proportion of a material's concentration in the octanol to that material's concentration in the aqueous portion. Lipophilic chemicals own a plus  $\log P$ , while hydrophilic chemicals own a minus  $\log P$ . According to Lipinski's rule,  $\log P$  should not overtake 5. Positive  $\log$  values were found for all of the triazole derivatives tested, indicating that they are lipophilic (B3, B4, and B5 at 1.8, 1.35, and 3.07, respectively). The total number of donor bonds (H-BD) in the nitrogen-hydrogen and oxygen-hydrogen media should be less than 5. The values 6,5,4 for the (B3,B4,B5) respectively Just as the total number of acceptor hydrogen bonds (H-BA) in an oxygen-nitrogen medium should not exceed 10, The values for compound B3,B4,B5. At (2,3,2) respectively An additional rule indicates that the rotatable bonds (RB) are less than 10 and values 5, 4 and 5 for compounds B3, B4, B5 respectively. Bioavailability refers to the mean and extent of accessibility of the active pharmaceutical ingredient (API) within the target site. The bioavailability of a compound, triazole B3-B5, ranged from 70% to 72%. Human intestinal absorption (HIA) is the one of the most pertinent lineaments of ADMET The HIA accounts for all of the triazoles compounds B3-B5 99.46%, 99.37%,99.57% : The aqueous solubility of compounds( $\log S$ )for compounds (B3,B4,B5)-3.264,-,2.805and-2.761 Classification of compounds can be according to solubility values ( $\log S$ ); Compounds with 0 and higher solubility value are highly soluble, those in the range of  $-2$  to  $0$  are very soluble, those in the range of  $-4$  to  $-2$  are soluble, those range of  $-6$  to  $-4$  moderately soluble, if less than  $-6$  are poorly soluble and insoluble if the value less than  $-10$ . Plasma

protein binding defines the degree of binding of a drug to plasma proteins .the values 78.9%,80.9% and87.2% for the compounds (B3,B4,B5)respectively It determines the degree of drug binding to plasma proteins. Protein binding can be measured in different ways, eg. Balanced dialysis (reference method), ultrafiltration, Ultra Centrifugation. The main plasma proteins that bind antibiotics are serum albumin and  $\alpha$ 1-acid glycoprotein. The term acute oral toxicity is often used in relation to the ability to kill and to determine the lethal dose, 50. The values ranged from 2.173, 2.063 and 1.921 for the values B3, B4, B5, respectively [35-41].

## Conclusion

A number of 1,2,4-triazole derivatives have been synthesized and produced in high yields. based on melting point range, infrared spectral values,  $^1\text{H}$  NMR, $^{13}\text{C}$ NMR and mass spectrometric data. It has been tested for antifungal efficacy against *Rhizoctonia solani* and *Fusarium* spp. when compared to the common pesticide beltanol and the computerized study In-silico for the five indicators: absorption, distribution, metabolism, excretion, and toxicity. Showed good indicators.

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